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## Key indicators

Single-crystal X-ray study  
 $T = 294$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.040  
 $wR$  factor = 0.112  
Data-to-parameter ratio = 16.1For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## Redetermination of 4-nitrobenzene-sulfonamide

In the crystal structure of the title compound,  $\text{C}_6\text{H}_6\text{N}_2\text{O}_4\text{S}$ , the nitro group is nearly coplanar with the aromatic ring, the dihedral angle being  $1.6(2)^\circ$ . The overall molecular organization is stabilized by well defined intermolecular hydrogen bonds, leading to the formation of an infinite planar network structure.

## Comment

It is known that the diuretically active title compound, (I), is an inhibitor of zinc-enzyme carbonic anhydrase which catalyses the reversible hydration of carbon dioxide (De Benedetti *et al.*, 1988). Its IR and NMR spectra have been reported previously (De Benedetti *et al.*, 1988; Pomerantz *et al.*, 1987; Dahn *et al.*, 1991). The crystal structure of (I) was originally analysed by Tremayne *et al.* (2002) from X-ray powder diffraction data using the Rietveld method. In addition, the crystal and molecular structure of the inclusion compound of the *cis-anti-cis* diastereomer of dicyclohexano-18-crown-6 with the title compound has been determined by X-ray analysis (Fonar *et al.*, 1992). In order to obtain more precise knowledge about the benzenesulfonamide compound, a single-crystal X-ray analysis of (I) was performed in our laboratory.

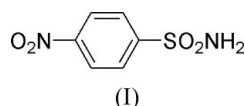


Fig. 1 shows a perspective view of the molecule, together with hydrogen-bonding connections to atoms in adjacent molecules. Angles  $\text{C1}-\text{S1}-\text{N1}$  [ $107.99(9)^\circ$ ],  $\text{C1}-\text{S1}-\text{O1}$

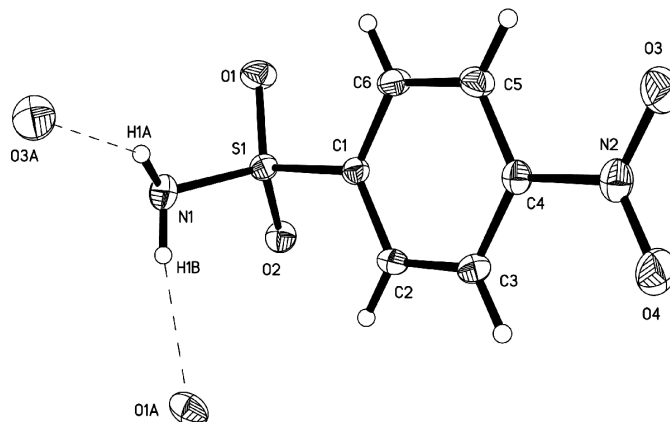
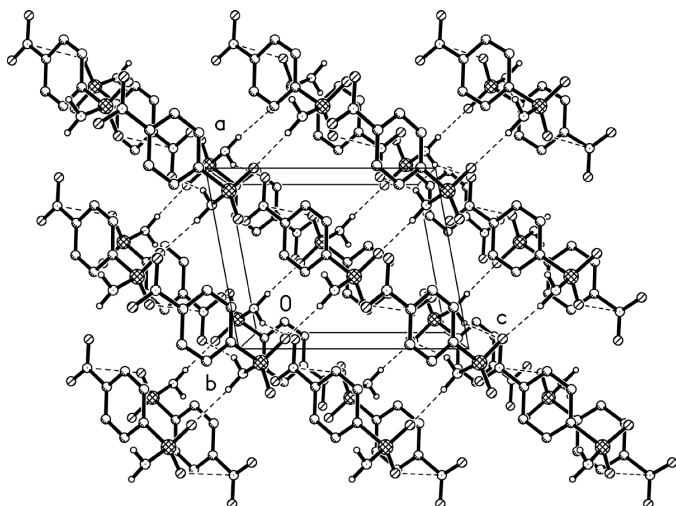


Figure 1

The molecular structure of (I), with displacement ellipsoids drawn at the 30% probability level. Dashed lines indicate hydrogen bonds. O1A and O3A correspond to O1<sup>ii</sup> and O3<sup>i</sup> in Table 1.



**Figure 2**  
Packing diagram for the title compound. Hydrogen bonds are shown as dashed lines.

[106.84 (9)°] and C1—S1—O2 [107.88 (9)°] are much narrower than O1—S1—O2 [119.75 (10)°]. This behaviour may be due to the mutual repulsion between the lone-pair electrons of two O atoms in the sulfonamide group. On the other hand, the nitro group is nearly coplanar with the aromatic ring, the dihedral angle being 1.6 (2)°. One of the sulfonamide H atoms approaches the nitro O atom of an adjacent molecule at a distance of 2.20 Å, while the other sulfonamide H atom approaches the sulfonamide O atom of another molecule at a distance of 2.18 Å, indicating the existence of two hydrogen bonds (Table 1), leading to the formation of an infinite planar network structure (Fig. 2).

## Experimental

The title compound was obtained from a commercial source (Aldrich). The crystal used for the data collection was obtained by slow evaporation of an ethanol solution at room temperature.

### Crystal data

$C_6H_6N_2O_4S$	$D_x = 1.620 \text{ Mg m}^{-3}$
$M_r = 202.19$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 3018 reflections
$a = 7.068 (1) \text{ \AA}$	$\theta = 1-27.5^\circ$
$b = 13.492 (2) \text{ \AA}$	$\mu = 0.37 \text{ mm}^{-1}$
$c = 8.841 (1) \text{ \AA}$	$T = 294 (2) \text{ K}$
$\beta = 100.459 (3)^\circ$	Block, colourless
$V = 829.1 (2) \text{ \AA}^3$	$0.50 \times 0.20 \times 0.18 \text{ mm}$
$Z = 4$	

### Data collection

Bruker SMART CCD area-detector diffractometer	1901 independent reflections
$\varphi$ and $\omega$ scans	1457 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.039$
$T_{\text{min}} = 0.835$ , $T_{\text{max}} = 0.936$	$\theta_{\text{max}} = 27.6^\circ$
5526 measured reflections	$h = -9 \rightarrow 8$
	$k = -17 \rightarrow 17$
	$l = -11 \rightarrow 9$

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0616P)^2 + 0.1102P]$
$R[F^2 > 2\sigma(F^2)] = 0.040$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.112$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.05$	$\Delta\rho_{\text{max}} = 0.25 \text{ e \AA}^{-3}$
1901 reflections	$\Delta\rho_{\text{min}} = -0.46 \text{ e \AA}^{-3}$
118 parameters	
H-atom parameters constrained	

**Table 1**

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N1—H1A $\cdots$ O3 <sup>i</sup>	0.90	2.20	3.025 (2)	153
N1—H1B $\cdots$ O1 <sup>ii</sup>	0.90	2.18	3.075 (3)	171

Symmetry codes: (i)  $\frac{1}{2} - x, y - \frac{1}{2}, \frac{3}{2} - z$ ; (ii)  $\frac{1}{2} + x, \frac{1}{2} - y, \frac{1}{2} + z$ .

H atoms were placed in geometrically calculated positions, with C—H = 0.93 and N—H = 0.90 Å, and were included in the final refinement in the riding-model approximation, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C or N})$ .

Data collection: SMART (Siemens, 1995); cell refinement: SAINT (Siemens, 1995); data reduction: SAINT and SHELXTL (Siemens, 1995); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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